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## WHAT IS CLAIMED IS:

- 1. A method for preparing a manganese compound for a lithium manganese complex oxide, comprising the step of simultaneously applying a mechanical force and a heat energy to a manganese compound to remove defects present in particles of said manganese compound, and to control the aggregation of micro particles and the shape of the aggregated particles.
- 2. The method for preparing the manganese compound according to claim 1, wherein a mechanical force and a heat energy are simultaneously applied to said manganese compound with adding one or more kinds of preparations selected from the group consisting of LiOH, LiOH H₂O, LiCH₃COO, LiOHO, LiCHO · H₂O, LiNO₃, and⟨a transition metal salt having a melting point of 200 ℃ or less.
- 3. The method for preparing the manganese compound according to claim 2, wherein the amount of said preparations is 0 to 20 wt% of the manganese compound.
- 4. The method for preparing the manganese compound according to claim 1, wherein said manganese compound is selected from the group consisting of electrolytic manganese dioxide (MnO<sub>2</sub>; EMD), chemical manganese dioxide (MnO<sub>2</sub>; CMD) Mn<sub>2</sub>O<sub>3</sub> and Mn<sub>3</sub>O<sub>4</sub>.
- 5. The method for preparing the manganese compound according to claim 2, wherein said manganese compound is selected from the group consisting of electrolytic manganese dioxide (MnO<sub>2</sub>; EMD), chemical manganese dioxide (MnO<sub>2</sub>; CMD), Mn<sub>2</sub>O<sub>3</sub> and Mn<sub>3</sub>O<sub>4</sub>
- 6. The method for preparing the manganese compound according to claim 1, wherein the applied mechanical force is 0.1 to 1000 dyne/cm², the range of the temperature of the applied heat energy is 50 to 200 °C, and the applied time is 5 minutes to 5 hours.
- 7. The method for preparing the manganese compound according to claim 2, wherein the applied mechanical force is 0.1 to 1000 dyne/cm², the range of the temperature of the applied heat energy is 50 to



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200  $\,^{\circ}$ C, and the applied time is 5 minutes to 5 hours.

8. The method for preparing the manganese compound according to claim 1, wherein a manganese compound having a shape without edge parts is prepared by using an angular shaped manganese compound as a raw material and applying mechanical force and heat energy.

- 9. The method for preparing the manganese compound according to claim 2, wherein a manganese compound having a shape? without edge parts is prepared by using an angular shaped manganese compound as a raw material and applying mechanical force and heat energy.
- 10. A method for preparing lithium manganese complex oxide with a spinel structure, comprising the steps of:
  - a) mixing
  - (i) a manganese compound prepared by the method comprising the step of simultaneously applying a mechanical force and a heat energy to a manganese compound to remove defects present in the particles of said manganese compound and to control the aggregation of micro particles and the shape of the aggregated particles; and
  - (ii) a lithium compound; and
  - b) calcining the mixture prepared in said step (a).
- 11. A method for preparing a lithium manganese complex oxide with a spinel structure according to claim 10, wherein the (ii) lithium compound of step (a) is selected from a lithium salt group consisting of LiOH, LiOH H<sub>2</sub>O, LiCH<sub>2</sub>COO, LiCHO, LiCHO H<sub>2</sub>O and LiNO<sub>3</sub>.
- 12. A method for preparing the lithium manganese complex oxide with a spinel structure according to claim 10, wherein the temperature of calcination of said step (b) is 400 to 900 °C, and the time of calcination is 1 to 30 hours.
- 13. A method for preparing the lithium manganese complex oxide with a spinel structure according to claim 11, wherein the temperature



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of calcination of said step (b) is 400 to 900  $^{\circ}$ C, and the time of calcination is 1 to 30 hours.

- 14. A lithium or lithium ion secondary battery comprising an anode, an electrolyte and a cathode using a lithium manganese complex oxide powder with a spinel structure as an active material, wherein said active material is a lithium manganese complex oxide with a spinel structure prepared by the method comprising the steps of:
- a) mixing

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- (i) a manganese compound prepared by the method comprising the step of simultaneously applying a mechanical force and a heat energy to a manganese compound to remove defects present in particles of the manganese compound and to control the aggregation of micro particles and the shapes of the aggregated particles; and
- (ii) a lithium compound; and
- b) calcining the mixture.

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